

SIMULTANEOUS DETERMINATION OF GLUCOSE, MALTOSE AND SUCROSE IN STARCH OF SIX VARIETIES OF CASSAVA

C. Carola C. Rojas^{1,3}, Baboo Nair², Adelina Herbas³, Björn Bergenståhl¹

¹Food Technology, Lund University, P.O Box 124, SE-221 00 Lund, Sweden

²Applied Nutrition, Lund University, P.O Box 124, SE-221 00 Lund, Sweden

³Centro de Alimentos y Productos Naturales, Universidad Mayor de San Simón, Cochabamba, Bolivia

Key words: Enzymatic hydrolysis, analysis, glucose, maltose, sucrose, Cassava. sugars determination is by using enzymatic

ABSTRACT

An enzymatic method has been developed to simultaneously determine glucose, maltose and sucrose in tubers. The principle of the method is based on an enzymatic cleavage of these disaccharides and specific measurement of the resulting glucose using an enzymatic procedure described by Holm (3).

The study comprises two parts; the first one includes modifications of the pretreatment method to minimize errors due to ethanol used to precipitate polymeric material interfering with the activity of α -glucosidase and invertase. In the second part the usability of the method has been probed by employing it on six samples of cassava that come from Chapare-Bolivia.

INTRODUCTION

Cassava (*Mannihot esculenta* Crantz) is a tropical root that has mainly carbohydrates in its composition such as starch, glucose, sucrose and fructose Huang (13). Lewthwaite (8) and Takata (11) have reported important changes in the composition of potato saccharides during its processing by showing variations in the sugars concentration which is an essential variable to determine the quality of the food. Reducing sugars has been identified as critical in potato for the tendency for acrylamide formation during heat processing of potato (14).

To determine the sugars content in plants and food, different techniques have been developed such as High Performance Liquid Chromatography (HPLC) (4,7); Gasses Chromatography (GC) (10). However, these techniques require considerable investments for the equipments and also normally demand a high cost of operation including large periods of time for the analysis. An alternative procedure for

methods, in this case the procedure is relatively simple and requires a UV-Vis spectrophotometer. The enzymatic kits are available for many sugars (1, 2, 9, 11, 5, 6) and also some other assays set have been developed for the determination of sugars (i.e. maltose, saccharose) by Dörner (5). Therefore, the main objective of this study was to develop a simple enzymatic method to determine the content of sugars (glucose, maltose and sucrose) in order to achieve a method with good performance in terms of accuracy-precision, easy-handling and simultaneous quick determination. The methodology is applied on a determination of the content of sugars in cassava of six varieties was characterized.

RESULTS AND DISCUSSION

To evaluate the ethanol interference, a screening with different concentrations of ethanol was performed to determine glucose as well as maltose and sucrose response obtained

Figures 1, 2 and 3 show the relationship between the ethanol concentration (in the final samples) and conversion (end concentration of the sugar). As shown in the Figure 1, the obtained glucose response varies as a function of the ethanol concentration. If compared the true values of glucose concentration (expected values) vs the obtained response (obtained values), it can be seen that the difference is not significant (less than 2%) up to 48% of ethanol in the final sample. As shown in the Figure 2, the obtained maltose response is more sensitive against the ethanol concentration. If compared the true values of maltose concentration (expected values) versus the obtained response (obtained values), it can be seen that the difference is not significant (less than 2%) up to 30% of ethanol in the final sample.

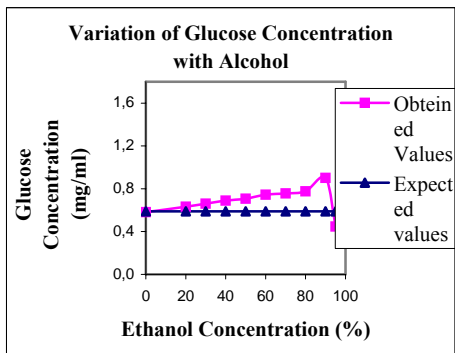


Figure 1. Influence of the ethanol concentration on the determination of glucose conversion

Figure 3 shows that there is a significant variation between the true values of sucrose (expected values) and the ones obtained by the analysis. This implies that there is a marked influence due to the presence of ethanol even when working at low concentrations.

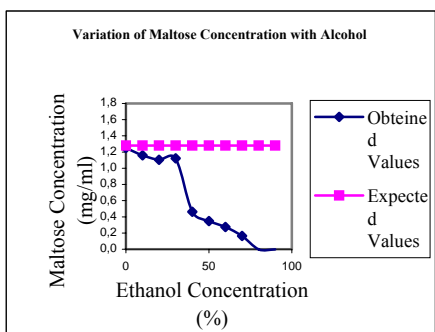


Figure 2. Influence of the ethanol concentration on the determination of maltose conversion

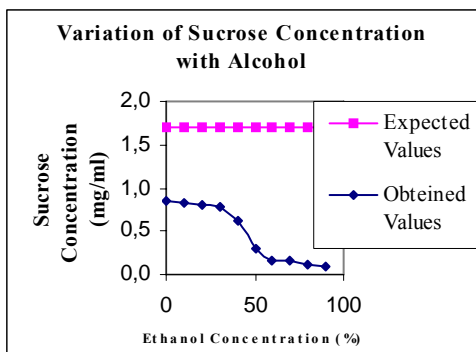


Figure 3. Influence of the ethanol concentration on the determination of sucrose conversion

Simultaneous analysis of glucose, maltose and sucrose in samples of cassava.

Once the technique was standardized, six varieties of Bolivian cassavas were analyzed; the results are

shown in Table 3.

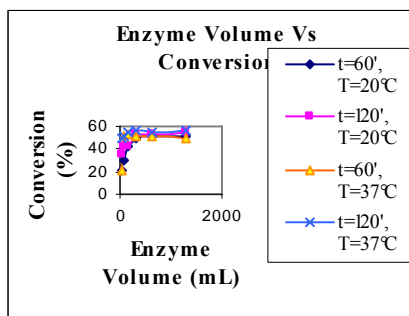


Figure 4. Sucrose conversion (mg glucose /mg sucrose*100) as function of time, temperature and volume of enzyme

Therefore, an experimental design (See Table 1) was performed to improve the sucrose conversion by modifying the conditions based on changes in incubation temperature, incubation time and initial concentration of the enzyme.

According to table 1, the appropriated conditions for the process to achieve best conversions were: incubation time = 60 minutes, incubation Temperature = 24 °C, V enzyme = 640 µL (150 U/ml), C ethanol = 30%)

Table 1

V enzima (µL)	T(°C)			
	24		37	
	t(min)			
40	60	120	60	120
80	XX	XX	XX	XX
160	XX	XX	XX	XX
320	XX	XX	XX	XX
640	XX	XX	XX	XX
1280	XX	XX	XX	XX

Determination of maltose and sucrose concentration

Based on the experiments and corrections, the sugar concentrations were calculated with the empiric deduced equations of Table 2.

The results show a high variability in terms of glucose for the six varieties, from 5 to 14 mg/g DM of sample. The content of sucrose varies between 36 and 46 mg/g DM and the content of maltose varies between 2 to 19 mg/g DM

Precision of the method

The precision of the method is analyzed from the data in Table 3. with 2 replicates, standard deviations and coefficients of variance were calculated for the six varieties of cassava. The standard deviation for glucose is 0.15 mg/g and about twice times higher for sucrose and maltose which is determined in two steps. The coefficient of error depends very much of the concentration (varies between 1 and 2%) and it is lowest for sucrose, present in a higher amount.

Method Specificity

The α -Glucosidase hydrolyzes the terminal, non-reducing 1,4-linked α -D-glucose residues with release of α -D-glucose (15)

EXPERIMENTAL

a) Substrates

The varieties of Cassava were: Bobore (a), Noveton (b), Valencia (c), Amarilla (d) and Cambayuca (e). Those come from "Estación Experimental la Jota", located in Chapare - Cochabamba -Bolivia The Bobore (f) sample comes from "Chimore. Experimental Plantations" IBTA-Chapare It is located in the Municipio de Chimoré of Chapare to 200 Km of the Cochabamba city). As a reference system potato starch p.a. was used. (Sigma-Aldrich)

b) Equipments

An analytical balance (Metler Toledo AE160), a water bath (Heto Birkeröd, Type 02 PG623, N° 7912, spectrophotometer (Perkin Elmer UV/VIS Lambda2).

c) Reagents

Reagents: anhydrous D-Glucose, Maltose Sucrose p.a. Buffer Solutions: Acetate Buffer 0,1 M, pH 6,6, Acetate Buffer 0.1 M, pH 4.6 and Buffer Tris 0.5 M. pH 7.

d) Enzymes

α -amylase (Fungamyl 800L) (Novo Nordisk, Copenhagen, Denmark), α -Glycosidase 240 U/ml (Sigma-Aldrich), Invertase 150 U/mg (Sigma-Aldrich) and Glucose Oxidase/oxidase (3), and Dihydrochloride like stain

f) Analytical Procedure.

The method basically consists of determinations of the glucose concentration after different steps of the hydrolysis. Glucose concentration was evaluated from samples that contain maltose by using α -glucosidase and sucrose by using

invertase. The background concentration of glucose was determined by an initial glucose analyses.

Standard Curve. Initially, 50 mg of anhydrous D-glucose was dissolved in 1000 mL of ethanol (48%). Dilutions were prepared ranging from 0.01-0.05 mg/mL to obtain the standard curve. 4 mL of glucose oxidase/oxidase enzyme solution was added then they were incubated for one hour at room temperature (20°C). The wavelength used for the determination was 450 nm.

Sample Preparation: 1 g of milled vacuum dried sample (particle size < 0.8 mm) was suspended in ethanol (46%) to make up 100 ml of suspension, the aim of the alcohol treatment is to precipitate all starch and proteins. The system was extracted in the shaker at room temperature during 40 min. Afterwards the suspension was filtrated by paper media filter 0.22 μ m.

Glucose Analysis: 1 mL of the filtered solution was mixed with 1 mL of buffert (pH 6) and 4 mL of glucose oxidase/oxidase (5,6 g/100 ml) as described by Holm (3). The mixture was incubated for one hour at room temperature and then the absorbance was read at 450 nm.

Maltose Analysis: 1 mL of the filtered solution was mixed with 1 mL of acetate buffer pH 6.6 and 0.2 mL of α -glucosidase (240 U/ml), the mix was incubated for one hour at room temperature according to the procedure by Gutmann (6). Afterwards, 0.1 mL of the obtained solution was mixed with 1.9 mL of water and 4 mL of glucose oxidase/oxidase (5.6 g/100 ml) as described. Holm (3). The incubation process was carried out for one hour at room temperature. The absorbance was read at 450nm.

Sucrose Analysis: 0.3 mL of the filtered solution was mixed with 1 mL of acetate buffer pH 4.6 and 0.640 mL of invertase (150U/ml), the mixture was incubated for one hour at room temperature; then, 0.5 mL of 0.5 M Tris-buffer pH 9 was added as described by Dömer (5). Afterwards, 0.6 mL of the obtained solution mixed with 1.4 ml of water and 4 mL of glucose oxidase/oxidase (5,6 g/100 ml) as described. Holm (3). The incubation process was carried out for one hour at room temperature. The absorbance was read at 450 nm.

Interference corrections of the ethanol in the determinations:

The samples contain 46% of ethanol during the

extraction of sugars. It can be assumed that the ethanol may inhibit the enzymatic reactions. Thus, the sensitivity of α -glucosidase, invertase and glucose oxidase against ethanol was tested.

Improvement on the sucrose conversion:

In order to improve the invertase hydrolysis an experimental design was carried to evaluate the conversion by taking into account the temperature. Incubation time and enzyme concentration are considered as the variables with more influence in the process.

CONCLUSION

The presented method allows for measurements of the content of free glucose, maltose and sucrose in starch rich foods such as cassava. The method is quick, easy and reliable. However, in the analyses, the starch and proteins need to be precipitated using ethanol. The study showed that the maximum concentration of ethanol to avoid interferences for determining free glucose, maltose and sucrose were 48%, 30%, 30% respectively. The conditions to perform the sucrose analysis need to be adjusted relative to previously published standard conditions (2).

ACKNOWLEDGEMENTS

The study was supported by Swedish International Development Agency (SIDA/SAREC) in the project "Desarrollo de nuevos productos y mejoramiento de las tecnologías tradicionales de procesamiento de raíces tropicales", between San Simon University (Bolivia) and Lund University (Sweden).

REFERENCES

- (1) BAUMANN VON G, G. K. "Die Bestimmung von Zuckern in Fruchtsäften - ein Vergleich der enzymatischen mit der Luff - Schoorl Methode." *Ind Obst Gemuserewert*, 1971,56, 165-170.
- (2) BERGMAYER, H. U. *Methoden der enzymatischen Analyse*. Ed. H. U. Bergmeyer, Dieter Oesterhelt *Angewandte Chemie International*, 1975, Verlag Chemie GmbH, Weinheim
- (3) HOLM, J., BJÖRCK, I., DREWS, A., & ASP, N.-G. A rapid method for analysis of starch. *Starch/Stärke*, 1986, 38, 224-226.
- (4) JACK N. DAVIES, RONALD J. KEMPTON "Changes in the individual sugars of tomato fruit during ripening." *J. Sci. Food Agric*, 1975, 26, 1103-1110.
- (5) Klaus M. Dörner "Quantitative Determination of lactose, Maltose, and Sucrose in Urine" *European Journal of Pediatrics* 1971, 26, 45-52.
- (6) GUTMANN, I. *Methods of Enzymatic Analysis*. Ed. HU Bergmeyer - New York: Academic Press, 1974.
- (7) HAILA K, K. J., HÄKKINEN U, TAHVONEN R.. "Sugar and organic acid contents of vegetables consumed in Finland during 1988-1989." *J Food Comp Anal* 1992, 5, 100-107.
- (8) LEWTHWAITE, S. L. S., K. H.; TRIGGS, C. M. "Free sugars composition of sweet potato cultivars after storage." *Crop Hortc. Sci.* 1997 25, 33-41.
- (9) MOELLERING H, G. W. "Determination of citrate with citrate lyase". *Anal Biochem*, 1996, 17, 369-376.
- (10) MOLNÁR-PERL I, M. M. "Rapid method for the simultaneous GC quantitation of acids and sugars in fruits and vegetables." *Food Add Contamin* 1992, 9, 505-514.
- (11) TAKAHATA, Y. N., T. SATO, T. "Changes in carbohydrates and enzyme activities of sweet potato lines during storage." *J. Agric. Food Chem.* 1995, 43, 1923-1928.
- (12) JOYCE S. VELTEROP, FEMKE VOS. "A Rapid and Inexpensive Microplate Assay for the Enzymatic Determination of Glucose, Fructose, Sucrose, L - Malate and Citrate in Tomato (*Lycopersicon esculentum*) Extracts and in Orange Juice." *Phytochemical Analysis*, 2001, 12, 299-304.
- (13) Y. H. HUANG, D. H. P., A. W. KILILI, AND C. E. JOHNSON. "Changes in invertase Activities and Reducing Sugar Content in Sweetpotato Stored at Different Temperatures" *Journal of Agricultural and Food Chemistry*, 1999, 47, 4927-4931
- (14) BECALSKI A, LAU BPY, LEWIS D,

SEAMAN SW, Acrylamide in Foods: Occurrence, Sources, and Modeling. J. Agricultural Food Chem, 2003, 51: 802-808

(15) MATSUSAKA, K.; CHIBA, S.; SHIMOMURA, T.; Purification and substrate specificity of brewer's yeast alpha-glucosidase, Agric. Biol. Chem., 1977, 41, 1917-1923

Table 2

Co = Free concentration of glucose, (without adding α -glucosidase or invertase)
 Cm = Glucose concentration after adding α -glucosidase
 Cs = Glucose concentration after adding invertase

$$C_{MALTLOSE} = (C_m - C_o)$$

$$C_{SUCROSE} = \frac{(C_s - C_m)}{0,53}$$

Table 3

Sample	Glucose Concentration (mg/g)	S.D Glucose (mg/g) n=2	C.V. (%) Glucose	Sucrose Concentration (mg/g)	S.D Sucrose (mg/g) n=2	C.V. (%) Sucrose	Maltose Concentration (mg/g)	S.D Maltose (mg/g) n=2	C.V. (%) Maltose
Cassava (a)	13,60	0,27	1,98	35,71	0,13	0,34	18,53	0,31	1,63
Cassava (b)	5,42	0,09	1,67	46,18	0,25	0,53	2,11	0,05	2,55
Cassava (c)	5,72	0,01	0,19	31,08	0,12	0,39	8,92	0,01	0,01
Cassava (d)	11,68	0,01	0,09	39,65	0,26	0,65	17,07	0,74	1,21
Cassava (e)	8,13	0,14	1,75	42,29	0,47	1,11	9,4	0,24	2,52
Cassava (f)	4,61	0,07	1,61	40,62	0,76	1,88	6,01	0,13	2,12
Standard deviation df =5		0,15	1,6		0,44	1,1		0,38	2,1