

MECHANISTIC VIEWS OF STEREOSELECTIVE SYNTHESIS OF TRI- AND TETRA-SUBSTITUTED ALKENES, PART I; THE ORGANIC CHEMISTRY NOTEBOOK SERIES, A DIDACTICAL APPROACH, Nº 3

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ABSTRACT

As underlined in two previous papers in: “The Organic Chemistry Notebook Series, a Didactical Approach”, the presentation of synthesis works in a verbal and graphical succinct manner, needs a didactical approach. Isomerically pure tri- and tetra-substituted alkenes are difficult to obtain as shown in several publications. We used a series of reactions to synthesize tri- and tetra-substituted alkenes as reviewed by W. Carruthers, and we have proposed didactical and mechanistic views for the reviewed reactions.

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ANALYSIS AND MECHANISTIC PROPOSALS

As academics we are concerned with the didactical importance of covering the needs of debutant students in organic synthesis. This is a third study in: “The Organic Chemistry Notebook Series, a Didactical Approach” [1,2]. The present article is an analytical and didactical approach to stereoselective synthesis of tri- and tetra-substituted alkenes as reviewed by W. Carruthers [3]. Tri- and tetra-substituted alkenes are difficult to obtain. Authors [3,4] signaled that the substrate α -chloro-aldehyde or -ketone, in its way to alkene, finds its critical step when reacting with Grignard reagent. Also, the most reactive conformation for the substrate is when the carbonyl group and the carbon-chlorine are antiparallel dipoles [3,4]. This conformation allows an addition of Grignard reagent very stereoselectively and from the side less hindered by R^1 and R^2 groups, on the α -carbon atom with respect to carbonyl [3,4]. The nucleophilic attacking group R^4 orients itself in an *anti* disposition with respect to the most hindering group between R^1 and R^2 [3,4]. The derived chlorohydrin is then submitted to stereoselective reactions to afford the alkene where three of the double bond substituents come from the α -chloro-aldehyde, -ketone and the fourth one comes from Grignard reagent [3,4]. Figure 1 shows the corresponding mechanistic view.

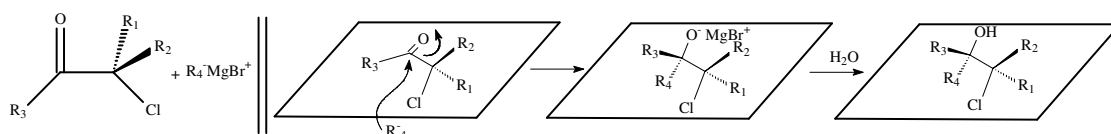


Figure 1. Halohydrin by nucleophilic attack (R^4) over $C=O$ of α -chloro-aldehyde, -ketone

As an example let us examine the synthesis of (*E*)-3-methyl-2-pentene [3,4] (Figure 2), and let us propose the corresponding mechanistic view (Figure 3).

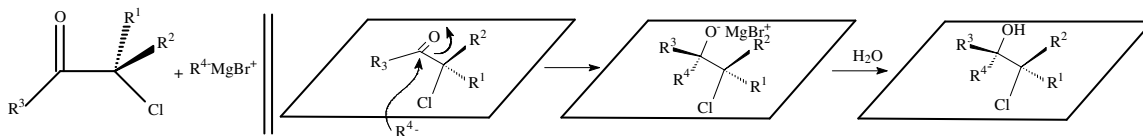


Figure 2. Synthesis of (*E*)-3-methyl-pentene as reviewed by W. Carruthers [3]

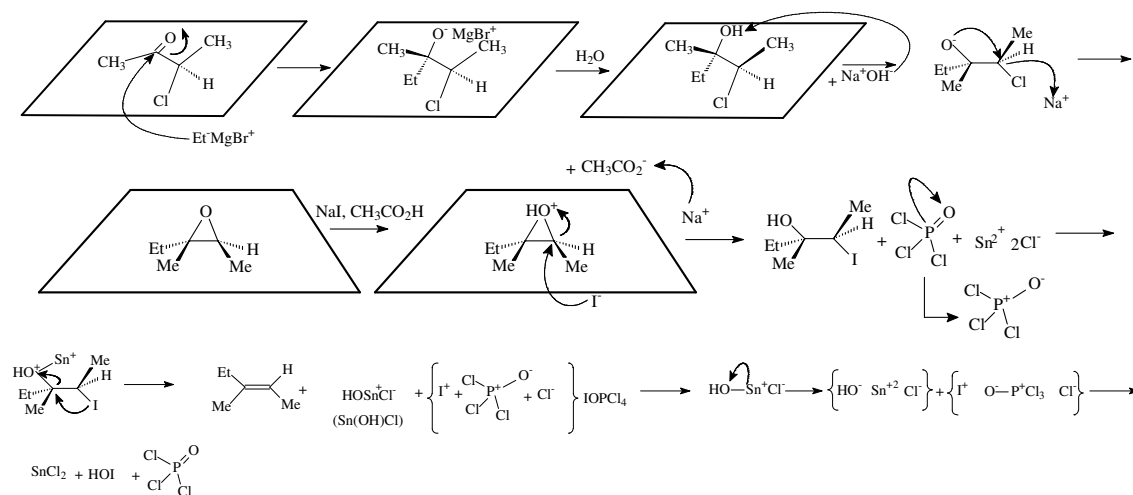


Figure 3. Synthesis of (E)-3-methyl-pentene, mechanistic view

Similarly, a series of stereo-selective reactions with methylmagnesium iodide (Grignard reagent) onto substrate 2-chloropentane-3-one gave rise to (Z)-3-methyl-2-pentene [3,4] as shown in the mechanism of Figure 4. These reaction series are an elegant way to obtain pure stereo-isomers.

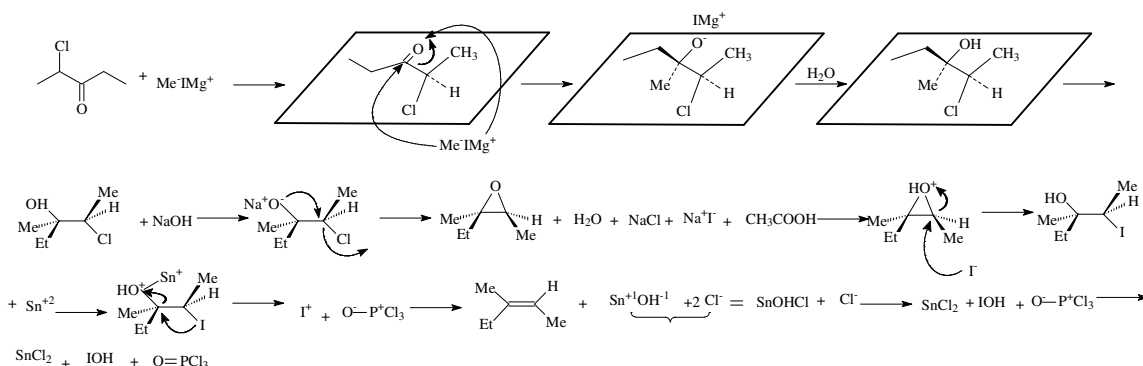


Figure 4. Synthesis of (Z)-3-methyl-2-pentene, mechanistic view

A different reaction to afford the stereo-specific obtaining of 2- or 3-alkylated allylic alcohol from a propargylic alcohol is achieved by reduction of propargylic alcohol into β - or γ -iodoallylic alcohols with aluminium hydride reagent (modified) and ulterior reaction of the reduction product with iodine [3,5], Figure 5.

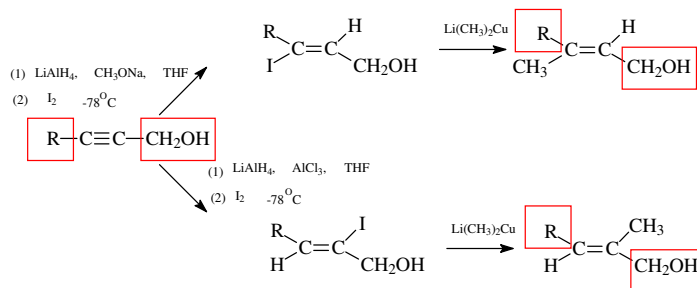


Figure 5. Synthesis of 3- or 2-alkylated allylic alcohols where the substituents, originally present in the propargyl alcohol, are trans to each other; reviewed by W. Carruthers [3]

Reduction is done with presence of NaOMe and the final product is only γ -iodoallylic alcohol. If reduction is carried out with LiAlH₄ in the presence of AlCl₃, final iodination gives final β -iodoallylic alcohol exclusively.

The resulting iodo compounds with LiR_2Cu afford the corresponding substituted allylic alcohol, where the original substituents of the propargyl alcohol, are now *trans* to each other (Figure 5) [3,5]. The explicit mechanism for these reactions are shown on Figure 6 and 7. Figure 6 shows the mechanism for the synthesis of β -iodinated, allylic alcohol.

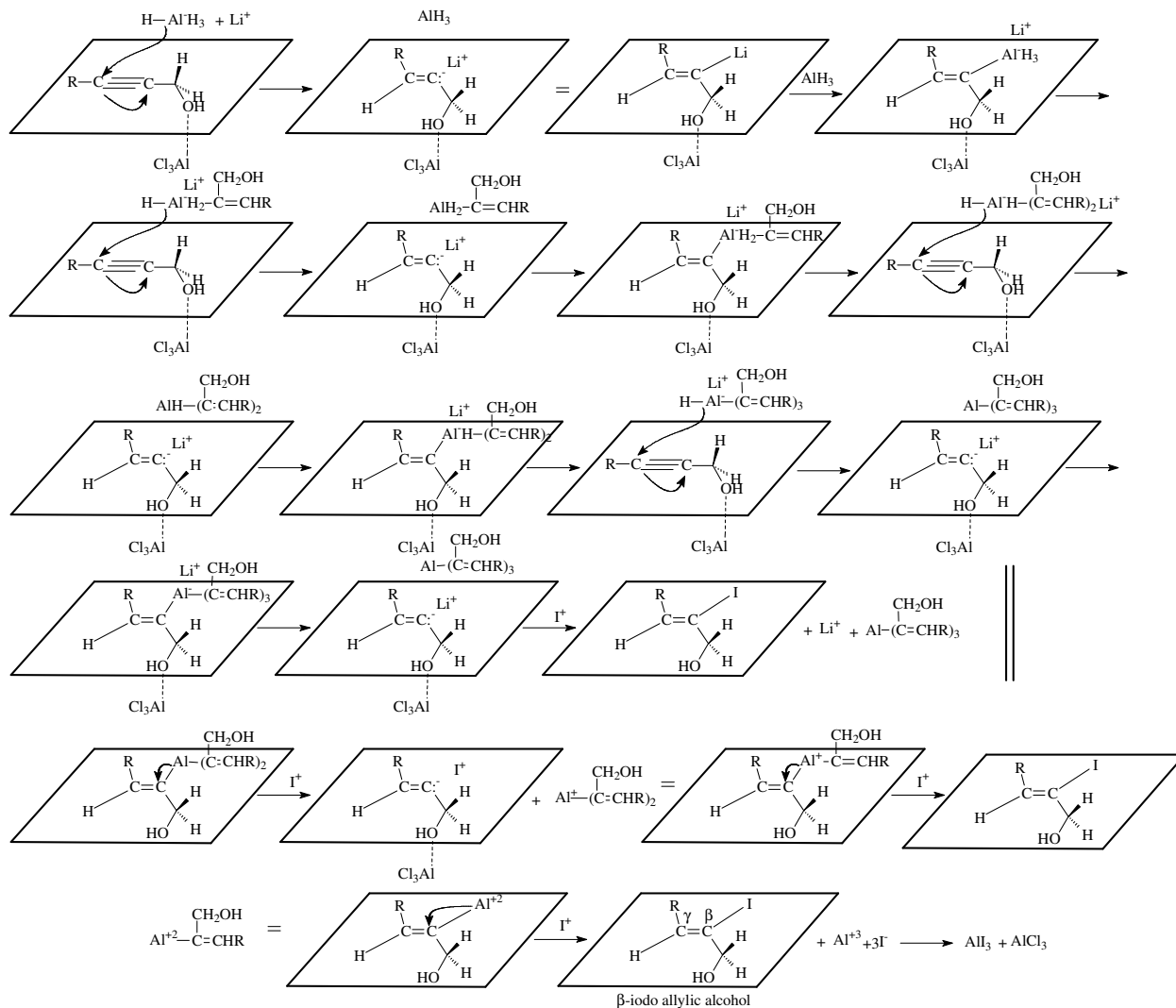


Figure 6. Synthesis of β -iodinated allylic alcohol, mechanistic view

For the γ -iodinated allylic alcohol, NaOCH_3 is employed (Figure 5) [3,5]. The corresponding mechanism is shown in Figure 7.

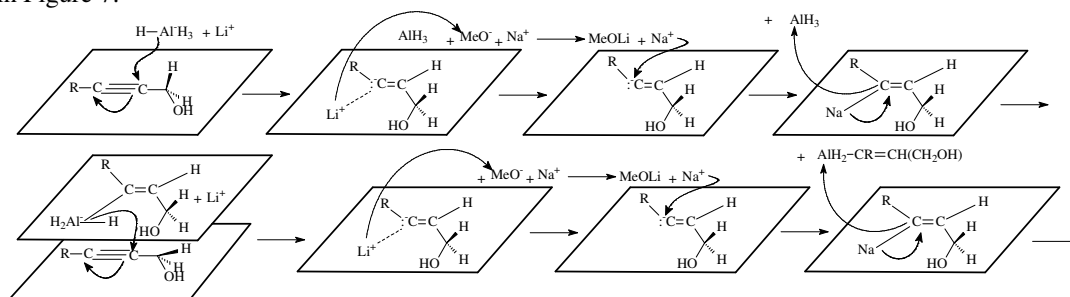


Figure 7. Synthesis of γ -iodinated allylic alcohol, mechanistic view

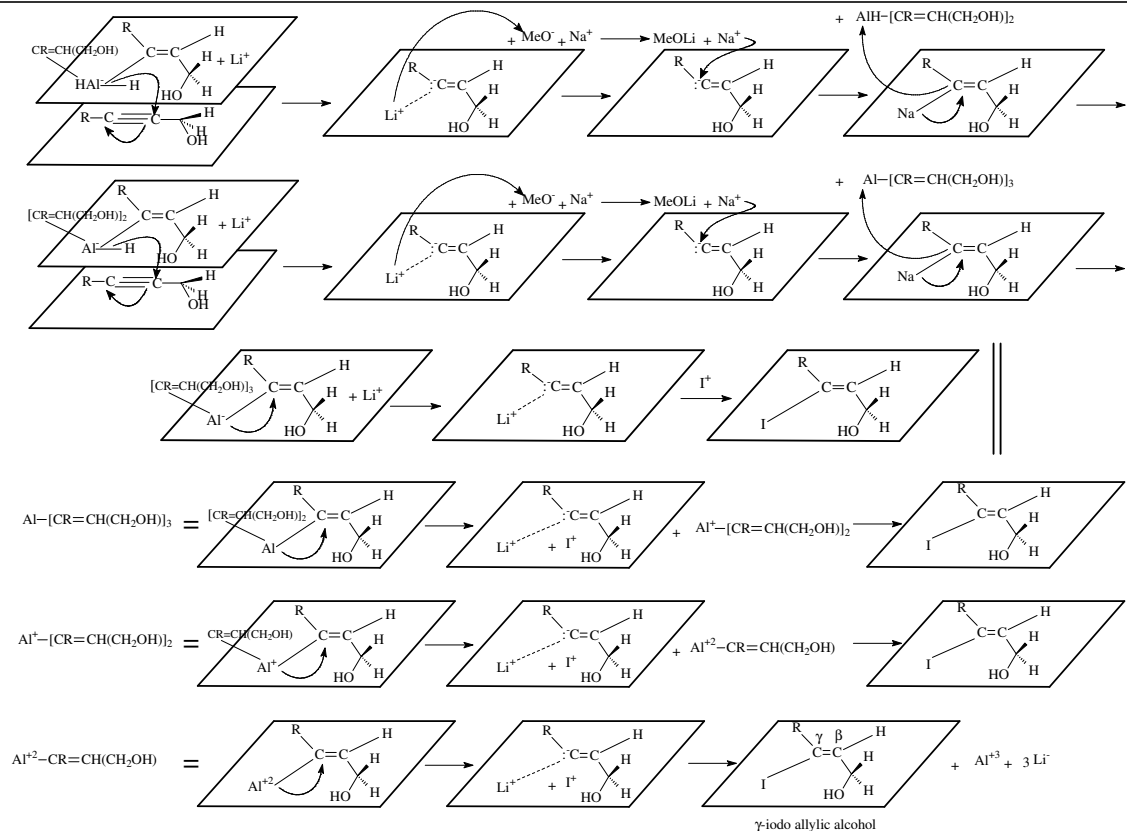


Figure 7. Synthesis of γ -iodinated allylic alcohol, mechanistic view (Cont.)

The γ -iodinated and β -iodinated allylic alcohols are easily transformed by action of lithium dimethylcuprate over substrates to afford 2-, 3-alkylated allylic alcohols as shown in Figure 8.

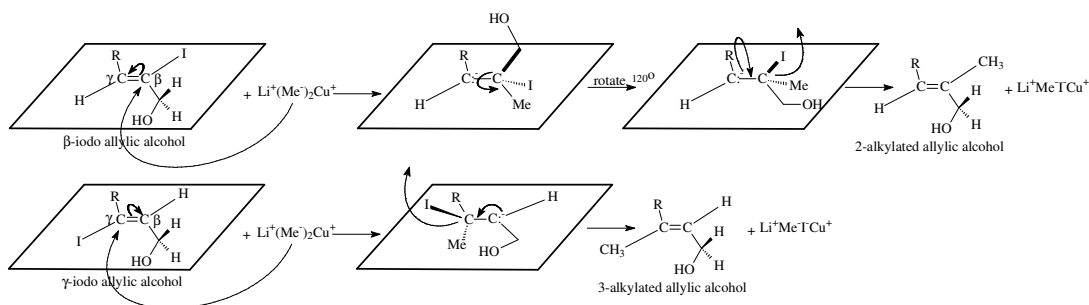


Figure 8. Transformation of β -, γ -iodinated allylic alcohols into 2-, 3-alkylated allylic alcohols; mechanisms

This method found application in the stereo-specific C=C bond formation in the synthesis of trisubstituted derivatives. This was the case for the key step in the synthesis of the *dl*-C18 *Cecropia* juvenile hormone, as reviewed by W. Carruthers [3,6], Figure 9.

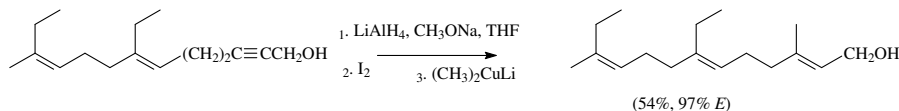


Figure 9. Key step in the synthesis of *dl*-C18 *Cecropia* juvenile hormone; reviewed by W. Carruthers [3]

Figure 10 is the mechanism corresponding to the key step in the synthesis of *dl*-C18 *Cecropia* juvenile hormone.

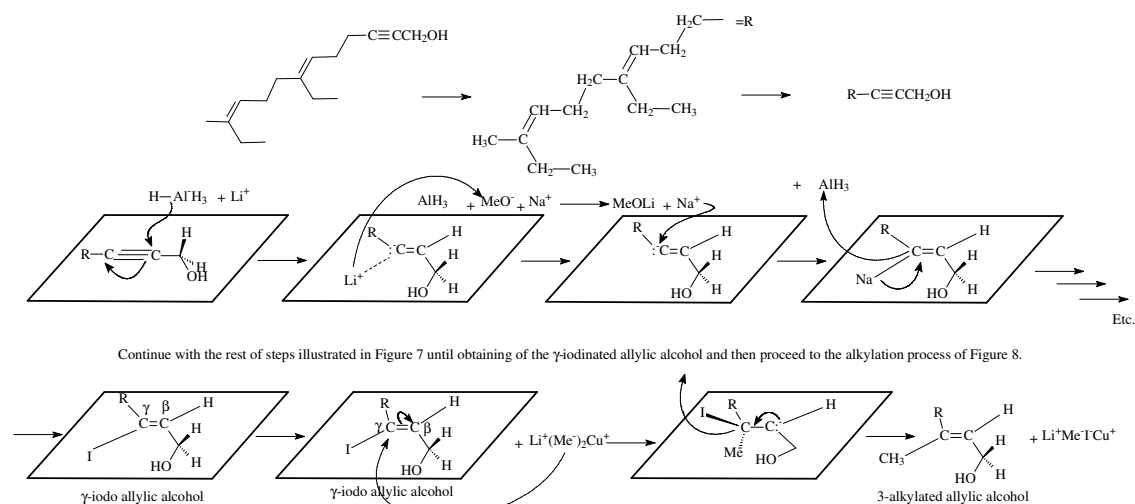


Figure 10. Key step in the synthesis of *dl*-C18 *Cecropia* juvenile hormone; mechanism

Organocopper and organoborane reagents are employed to obtain stereoselectively tri- and tetra-substituted alkenes by addition on alkynes [3,7]. Organocuprates lead to $\beta\beta$ -dialkylacrylic esters by reaction with $\alpha\beta$ -acetylenic esters (Figure 11) [3]. The structure of each stereoisomer depends on the temperature and the solvent [3]. High yield of $\beta\beta$ -dialkylacrylic ester is achieved at -78°C in THF [3]. Contrasting with reactions employing propargyl alcohols, this reaction produces alkenes in which the substituents in the acetylenic precursor are *cis* to each other in the alkene [3].

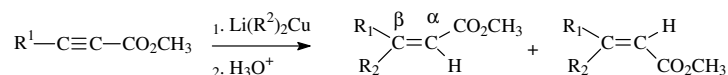


Figure 11. Trisubstituted alkenes stereoselectively obtained from alkynes by addition of organocuprates; reviewed by W. Carruthers [3]

The mechanistic views conducting to alkenes by this method are exposed in Figure 12.

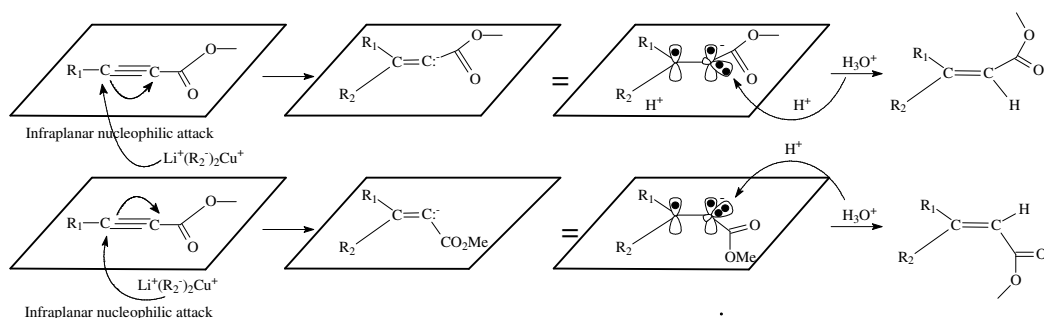


Figure 12. Trisubstituted alkenes stereoselectively obtained from alkynes by addition of organocuprates; mechanism

Organocopper(I) reagents add readily to terminal alkynes giving rise to 1-alkenylcopper(I) compounds [3]. These organocopper(I) reagents are RCu . Alkyl copper(I) compounds that can be obtained from Grignard reagents and an equimolar quantity of copper(I) bromide or copper(I) bromide-dimethylsulphide complex [3]. Copper adds to the alkyne on the terminal carbon, adding the alkyl group of the alkylcopper(I) reagent, in a *syn* manner [3]. This kind of products, alkenylcopper(I) compounds, react with electrophiles like alkyl halides, $\alpha\beta$ -unsaturated ketones and epoxides to afford trisubstituted alkenes with almost complete retention of configuration [3,7,8]. See Figure 13. The elaborated mechanism is shown in Figure 14.

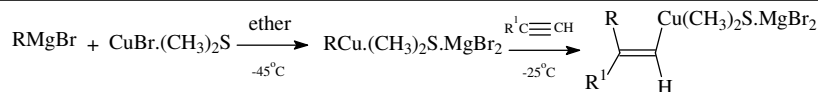


Figure 13. Trisubstituted alkenes stereoselectively obtained from alkynes by addition of organocopper(I) reagents; reviewed by W. Carruthers [3]

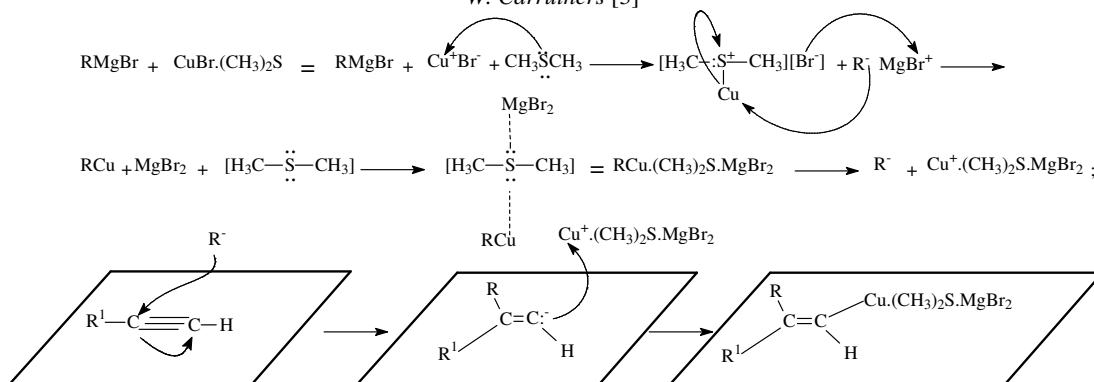


Figure 14. Trisubstituted alkenes stereoselectively obtained from alkynes by addition of organocopper(I) reagents; mechanistic view

The method was employed as shown in the examples of Figure 15.

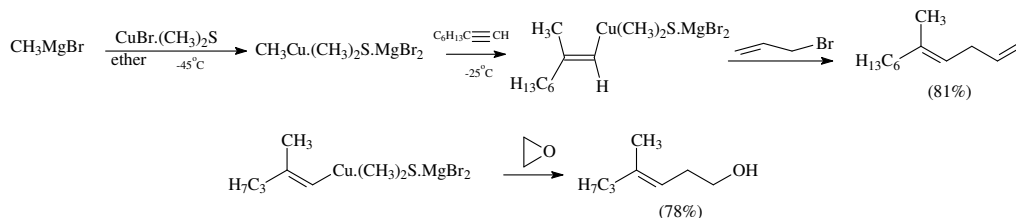


Figure 15. Trisubstituted alkenes stereoselectively obtained from alkynes by addition of organocopper(I) reagents, other examples; reviewed by W. Carruthers [3]

The corresponding mechanistic views are exposed in Figures 16 and 17.

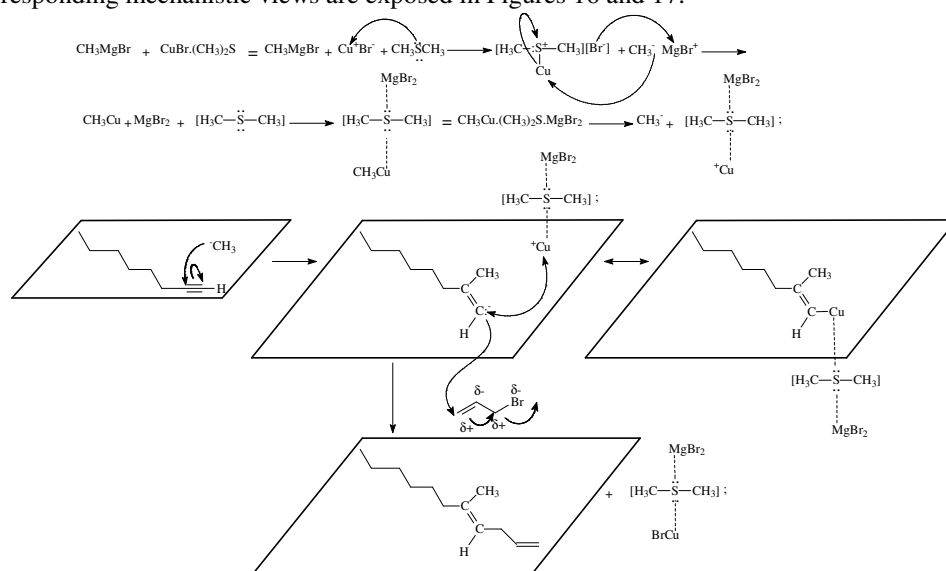


Figure 16. Trisubstituted alkenes stereoselectively obtained from alkynes by addition of organocopper(I) reagents, other examples; mechanistic views

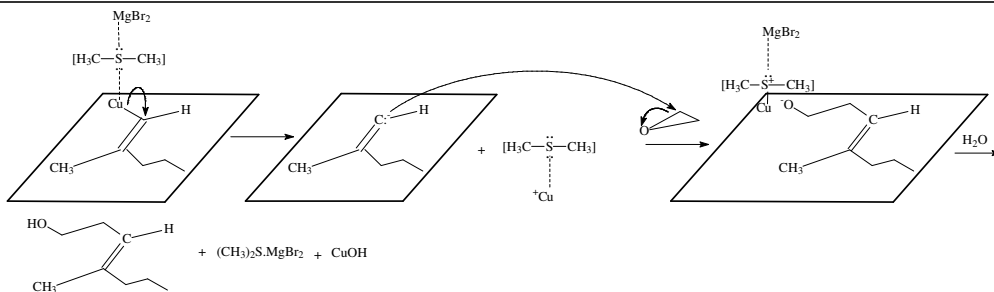


Figure 17. Trisubstituted alkenes stereoselectively obtained from alkynes by addition of organocopper(I) reagents, other examples; mechanistic views

Alkenyl iodides also react in the presence of $\text{Pd}(\text{PPh}_3)_4$ as catalyst to give conjugated dienes [3,9], Figure 18.

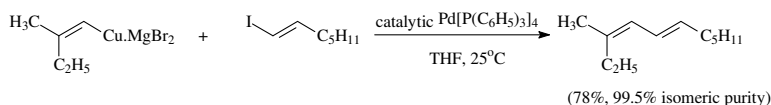


Figure 18. Alkenyl iodide catalyzed by $\text{Pd}[\text{P}(\text{C}_6\text{H}_5)_3]_4$ to afford conjugated diene; reviewed by W, Carruthers [3]

The corresponding mechanism is exposed in Figure 19.

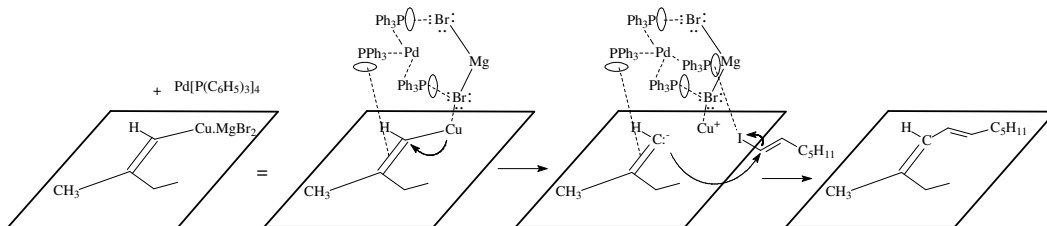


Figure 19. Alkenyl iodide catalyzed by $\text{Pd}[\text{P}(\text{C}_6\text{H}_5)_3]_4$ to afford conjugated diene; mechanism

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